

Synthesis of 2-Substituted 3-Aroylindenes via Palladium-Catalyzed Carbonylative Cyclization of Diethyl 2-(2-(1-Alkynyl)phenyl)malonates with Aryl Halides

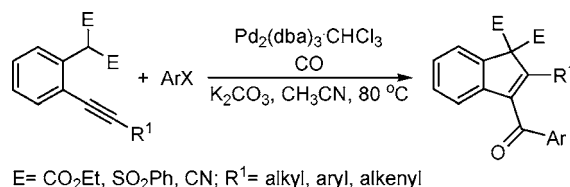
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ABSTRACT



The palladium-catalyzed reaction of readily accessible diethyl 2-(2-(1-alkynyl)phenyl)malonates with aryl halides under a balloon pressure of CO produced 2-substituted 3-arylyndenes in good yields. The reaction is believed to proceed via cyclization of the alkyne containing a proximate nucleophilic center promoted by an acylpalladium complex.

The transition metal-catalyzed cyclization of alkynes has recently proven to be a powerful method for the construction of a variety of carbo- and heterocycles.¹ Among them, the palladium-catalyzed carbonylative cyclization of alkynes bearing proximate nucleophilic centers with use of unsaturated organic halides/triflates in the presence of carbon monoxide has also been shown to be extremely effective for the synthesis of ketone-containing isoquinolines,² indoles,³

and benzofurans.⁴ Such reactions could generate one carbon–heteroatom bond and two carbon–carbon bonds in a single synthetic operation. However, the corresponding reactions with carbon nucleophiles to give three carbon–carbon bonded species are rare.

Very recently, we described a convenient approach to the synthesis of 2,3-disubstituted indenes from diethyl 2-(2-(1-alkynyl)phenyl)malonates and a wide variety of organic halides (Scheme 1).⁵ Our continuing interest in the synthesis of indenes derivatives by this chemistry as well as the valuable biological activity⁶ exhibited by some substituted indenes (their metallocene complexes may also be utilized to catalyze olefin polymerization)⁷ led us to investigate the extension of the methodology to the synthesis of substituted

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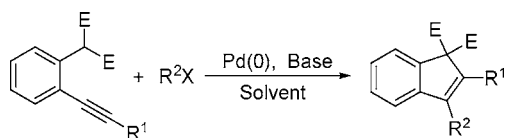
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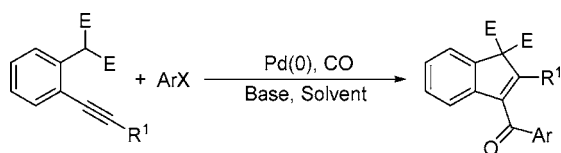
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Scheme 1



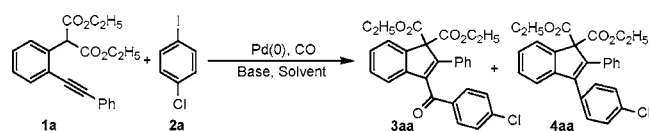
indenes. Herein, we wish to report a successful synthesis of 2-substituted 3-aryloindenes by the palladium-catalyzed carbonylative cyclization of diethyl 2-(2-(1-alkynyl)phenyl)malonates with aryl halides in the presence of CO (Scheme 2).

Scheme 2



Initially, we started out our investigation of the reaction by using 1.0 equiv of diethyl 2-(2-(2-phenylethynyl)phenyl)malonate (**1a**; 0.2 mmol), 1.2 equiv of 4-chloriodobenzene under a balloon pressure of CO employing 5 mol % of Pd(PPh₃)₄ and 2.0 equiv of K₂CO₃ in DMF at 100 °C for 24 h, reaction conditions that were used in our earlier palladium-catalyzed synthesis of 2,3-disubstituted indenes.⁵ The desired product, diethyl 3-(4-chlorobenzoyl)-2-phenyl-1*H*-indene-1,1-dicarboxylate (**3aa**), was formed in only a 34% isolated

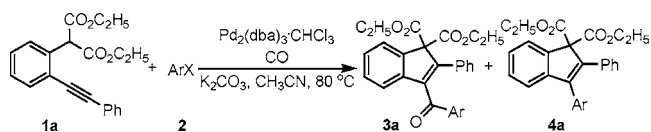
Table 1. Optimization of the Palladium-Catalyzed Carbonylative Cyclization of Diethyl 2-(2-(2-Phenylethynyl)phenyl)malonate and 4-Chloriodobenzene^a



entry	palladium catalysis	base	solvent	temp (°C)	isolated yield of 3aa (%) ^b
1	Pd(PPh ₃) ₄	K ₂ CO ₃	DMF	100	34 (52)
2	Pd(PPh ₃) ₄	(<i>n</i> -Bu) ₃ N	DMF	100	no reaction
3	Pd(PPh ₃) ₄	KOAc	DMF	100	no reaction
4	Pd(PPh ₃) ₄	KO <i>t</i> -Bu	DMF	100	0 (65)
5	Pd(PPh ₃) ₄	K ₂ CO ₃	DMSO	100	20 (35)
6	Pd(PPh ₃) ₄	K ₂ CO ₃	CH ₃ CN	80	22 (0)
7	Pd ₂ (dba) ₃ ·CHCl ₃	K ₂ CO ₃	CH ₃ CN	80	83 (0)
8	Pd(dba) ₂	K ₂ CO ₃	CH ₃ CN	80	55 (0)
9	Pd(OAc) ₂ /PPh ₃	K ₂ CO ₃	CH ₃ CN	80	15 (0)

^a Reactions were carried out on a 0.2 mmol scale in 2.0 mL of solvent under CO at 100 °C or 80 °C with 1.0 equiv of **1a**, 1.2 equiv of **2a**, 2.0 equiv of base, and 0.05 equiv of [Pd]. ^b The numbers in parentheses are the isolated yields of the corresponding 2-phenyl 3-aryloindene **4aa**.

Table 2. Palladium-Catalyzed Carbonylative Cyclization of Diethyl 2-(2-(2-Phenylethynyl)phenyl)malonate with Aryl Halides^a



entry	ArX (2)	time (h)	Ar (3a)	isolated yield (%) ^b
1	<i>p</i> -ClC ₆ H ₄ I	24	<i>p</i> -ClC ₆ H ₄ (3aa)	83 (0)
2	<i>p</i> -CH ₃ O ₂ CC ₆ H ₄ I	24	<i>p</i> -CH ₃ O ₂ CC ₆ H ₄ (3ab)	90 (0)
3	<i>p</i> -NO ₂ C ₆ H ₄ I	24	<i>p</i> -NO ₂ C ₆ H ₄ (3ac)	75 (0)
4	<i>p</i> -CF ₃ C ₆ H ₄ I	24	<i>p</i> -CF ₃ C ₆ H ₄ (3ad)	83 (0)
5	<i>p</i> -CH ₃ C ₆ H ₄ I	48	<i>p</i> -CH ₃ C ₆ H ₄ (3ae)	52 (7)
6	<i>p</i> -CH ₃ OC ₆ H ₄ I	48	<i>p</i> -CH ₃ OC ₆ H ₄ (3af)	42 (6)
7	PhI	24	Ph (3ag)	61 (0)
8	<i>m</i> -CH ₃ C ₆ H ₄ I	48	<i>m</i> -CH ₃ C ₆ H ₄ (3ah)	51 (0)
9	<i>m</i> -ClC ₆ H ₄ I	24	<i>m</i> -ClC ₆ H ₄ (3ai)	89 (0)
10	<i>m</i> -NO ₂ C ₆ H ₄ I	24	<i>m</i> -NO ₂ C ₆ H ₄ (3aj)	85 (0)
11	<i>o</i> -CH ₃ C ₆ H ₄ I	48	<i>o</i> -CH ₃ C ₆ H ₄ (3ak)	62 (0)
12	<i>o</i> -CH ₃ O ₂ CC ₆ H ₄ I	24	<i>o</i> -CH ₃ O ₂ CC ₆ H ₄ (3al)	78 (15)
13	2-iodothiophene	48	2-thienyl (3am)	65 (0)
14	<i>p</i> -NO ₂ C ₆ H ₄ Br	48	<i>p</i> -NO ₂ C ₆ H ₄ (3ac)	15 (22)

^a All reactions were carried out under the optimal conditions reported in the text. ^b The numbers in parentheses are the isolated yields of the corresponding 2-phenyl 3-aryloindenes **4a**.

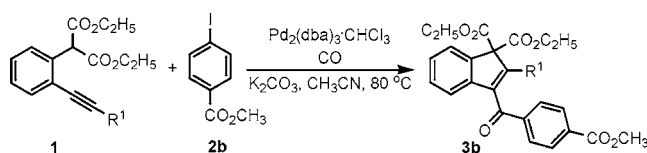
yield, the other diethyl 3-(4-chlorophenyl)-2-phenyl-1*H*-indene-1,1-dicarboxylate (**4aa**) was also isolated in 52% yield (Table 1, entry 1). 2-Phenyl 3-aryloindene **4aa** was formed without incorporation of CO by a process reported previously by us.⁵ Tri-*n*-butylamine and KOAc were ineffective (Table 1, entries 2 and 3). When KO*t*-Bu was used as the base, only the 2-phenyl 3-aryloindene **4aa** was isolated (Table 1, entry 4).

Changing the solvent from DMF to DMSO did not enhance the yield of **3aa** or the selectivity between the two indene products (Table 1, entry 5). Using acetonitrile as the solvent at 80 °C led to cleaner reaction, affording the desired product **3aa** in 22% yield with none of the side product **4aa** (Table 1, entry 6). Fortunately, the yield increased to 83% when the catalyst Pd₂(dba)₃·CHCl₃ was employed instead of Pd(PPh₃)₄ (Table 1, entry 7). Other palladium catalysts tested, such as Pd(dba)₂ and Pd(OAc)₂/PPh₃, were less effective (Table 1, entries 8 and 9). The optimum reaction conditions thus far developed employ 1.0 equiv of **1a**, 1.2 equiv of the aryl halide, 5 mol % of Pd₂(dba)₃·CHCl₃, and 2.0 equiv of K₂CO₃ in CH₃CN at 80 °C under a balloon pressure of CO.

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Table 3. Palladium-Catalyzed Carbonylative Cyclization of Diethyl 2-(2-(1-Alkynyl)phenyl)malonates with Methyl 4-Iodobenzoate^a

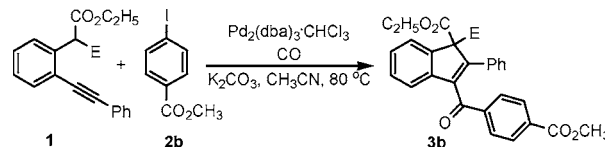


entry	R ¹	time (h)	3b	isolated yield (%) ^b
1	<i>p</i> -CH ₃ OC ₆ H ₄ (1b)	24	3bb	80 (0)
2	<i>p</i> -NO ₂ C ₆ H ₄ (1c)	24	3cb	93 (0)
3	<i>p</i> -BrC ₆ H ₄ (1d)	24	3db	85 (0)
4	1-cyclohexenyl (1e)	24	3eb	70 (0)
5	<i>n</i> -pentyl (1f)	24	3fb	60 (0)

^a All reactions were carried out under the optimal conditions reported in the text. ^b The numbers in parentheses are the isolated yields of the corresponding 2-substituted 3-arylidenes.

To extend the general applicability of this carbonylative cyclization reaction, the reaction of diethyl malonate alkyne **1a** with various aryl halides was carried out under the above-optimized conditions, and the results are summarized in Table 2. Aryl iodides bearing an electron-withdrawing group or an electron-donating group in the para, meta, and ortho positions afforded the corresponding multiply substituted indenenes **3a** in moderate to good yields (Table 2, entries 1–6 and 8–12). Aryl iodides bearing an electron-withdrawing group in the para position usually led to good yields of the 2-phenyl 3-arylidenes (Table 2, entries 1–4). When 4-methyliodobenzene and 4-iodoanisole were employed in the reaction with substrate **1a** (Table 2, entries 5 and 6), the corresponding products **3ae** and **3af** were isolated in 52% and 42% yields along with low yields of the side products diethyl 2-phenyl-3-*p*-tolyl-1*H*-indene-1,1-dicarboxylate (**4ae**) and diethyl 3-(4-methoxyphenyl)-2-phenyl-1*H*-indene-1,1-dicarboxylate (**4af**). Iodobenzene also afforded **3ag** in satisfactory yield (Table 2, entry 7). The reactions of **1a** and aryl iodides with an electron-withdrawing group, such as a Cl or NO₂ group in the meta position, afforded the desired products **3ai** and **3aj** in high yields (Table 2, entries 9 and 10). 3-Methyliodobenzene gave a modest yield of the indene product **3ah** (Table 2, entry 8). Aryl iodides bearing an electron-withdrawing group or an electron-donating group in the ortho position have also proven successful. For example, the reaction of 2-methyliodobenzene and methyl 2-iodobenzoate produced 62% and 78% yields of the desired products, respectively (Table 2, entries 11 and 12). In addition, the use of 2-iodothiophene also afforded the indene **3am** in 65% yield (Table 2, entry 13). In most of these reactions, only a very small amount of or no 2-phenyl-3-arylidene **4a** was isolated. Thus, these reactions exhibit good reaction selectivities. Unlike aryl iodides, aryl bromides such as 4-bromonitrobenzene gave the low yield of diethyl 3-(4-nitrobenzoyl)-2-phenyl-1*H*-indene-1,1-dicarboxylate (**3ac**) and poor selectivity (Table 2, entry 14).

Table 4. Palladium-Catalyzed Carbonylative Cyclization of Ethyl Acetate Alkynes with Methyl 4-Iodobenzoate^a



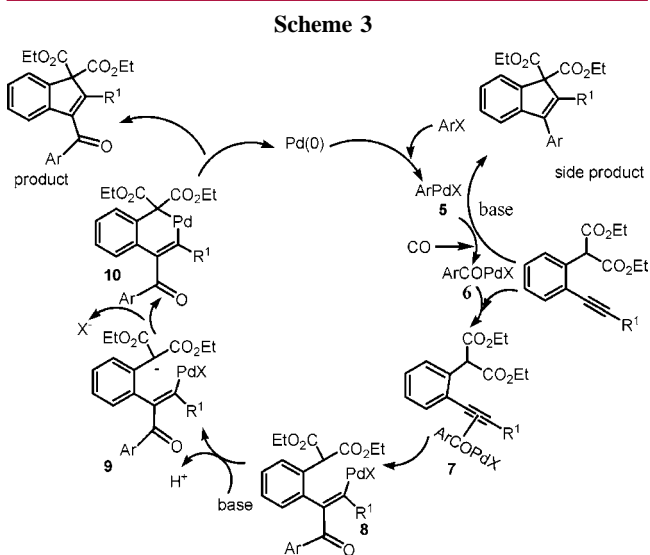
entry	E	time (h)	3b	isolated yield (%) ^b
1	SO ₂ Ph (1g)	24	3gb	55 (0)
2	CN (1h)	24	3hb	65 (0)

^a All reactions were carried out under the optimal conditions reported in the text. ^b The numbers in parentheses are the isolated yields of the corresponding 2-phenyl 3-arylidenes.

We have also investigated the reactions of diethyl malonate alkynes containing different R¹ groups at the end of the triple bond with an aryl iodide. With methyl 4-iodobenzoate, diethyl malonate alkynes with an electron-withdrawing group or an electron-donating group on the benzene ring were employed in the reaction, and the corresponding products were isolated in high yields (Table 3, entries 1–3). Diethyl malonate alkyne **1e** bearing a 1-cyclohexenyl group afforded the corresponding 2-substituted 3-arylidene **3eb** in a good yield of 70% (Table 3, entry 4). Diethyl malonate alkyne **1f** containing an *n*-pentyl group afforded the desired product **3fb** in a 60% yield (Table 3, entry 5).

Furthermore, ethyl acetate alkynes with different electron-withdrawing groups, such as ethyl 2-(2-(2-phenylethynyl)-phenyl)-2-(phenylsulfonyl)acetate (**1g**) and ethyl 2-cyano-2-(2-(2-phenylethynyl)phenyl)acetate (**1h**), have also been allowed to react with methyl 4-iodobenzoate to afford moderate yields of the desired products **3gb** and **3hb** (Table 4, entries 1 and 2).

The mechanism shown in Scheme 3 is proposed for this process. It is similar to mechanisms proposed in previously



reported palladium-catalyzed syntheses of isoquinolines,² indoles,³ benzofurans,⁴ and furans.⁸ It consists of the following key steps: (1) oxidative addition of the aryl halide to the Pd(0) catalyst, followed by CO insertion,⁹ (2) coordination of the resulting acylpalladium intermediate **6** to the alkyne triple bond to form complex **7**,^{2,3,4,8} (3) insertion of the alkyne in the aryl palladium complex **7** to form a vinylic palladium intermediate, **8**, (4) generation of a carbanion by the base, (5) intramolecular nucleophilic attack of the carbanion on the vinylic palladium intermediate to afford a palladacyclic intermediate, **10**,¹⁰ and (6) reductive elimination of the intermediate to furnish the indene and regenerate the Pd(0) catalyst. One competing process is cyclization of the starting material diethyl malonate alkynes

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promoted by an arylpalladium intermediate **5** to afford 2-substituted 3-arylindenes.⁵

In summary, an efficient, palladium-catalyzed synthesis of 2-substituted 3-arylindenes has been developed. A variety of aryl iodides undergo this process, giving the desired products in moderate to good yields with good reaction selectivities. In particular, three carbon-carbon bonds are formed in a single operative step.

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Supporting Information Available: Typical experimental procedure and characterization for all products. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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